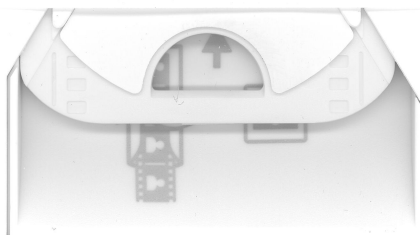
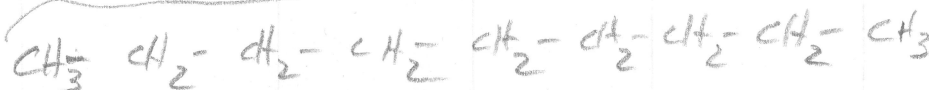


Formula	Name	Type (Ionic or Covalent)	# of Valence Electrons	Lewis Dot Diagram	Molecular Shape (Linear, Bent , or Tetrahedral)
$2 + 6 = 8$ H_2O	Water	covalent	8	$H - \ddot{O} - H$	Bent
Br_2	Bromine	covalent	14	$\ddot{Br} - \ddot{Br} :$	linear
$1 + 7$ HF	Hydrofluoric Acid	covalent	8	$H - \ddot{F} :$	linear
$36 + 20$ C_9H_{20}	nonane	covalent	56		tetrahedral



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